cooled, treated with EtOAc (10 ml), and refluxed for 30 min. After addn of $\rm H_2O$ and 1 N NaOH alternately, the mixt was filtered and concd in vacuo to remove THF. The residue was acidified with 6 N HCl and extd with Et₂O. The aq acid soln was basified and extd with Et₂O, and the dried Et₂O ext was treated with ethanolic HCl to afford 0.51 g (yield, 78%) of $8 \cdot \rm HCl$.

(+)-3-Hydroxymorphinan (12). A CHCl₃ soln (25 ml) contg 4.0 g (0.0015 mole) of 7 was treated with ethyl chloroformate (20 ml) and anhyd K CO₃ (2 g). The reaction mixt was stirred for 24 hr and the CHCl₃ then removed in vacuo. The residue was suspended in 1 N HCl (50 ml) and extd with Et₂O. Removal of Et₂O afforded an oil (2.5 g) which was dissolved in a mixt of glacial AcOH (10 ml) and 48% HBr (10 ml). After refluxing under N₂ for 6 hr, the reaction mixt was poured in ice water (300 ml), extd with Et₂O, and basified to afford 1.1 g (31%) of 12, mp 260-263° [reported for (-)-3-hydroxymorphinan, mp 260-262°]. 18

Norcodeine (7). A soln of 3.8 g (0.012 mole) of codeine (4) in CHCl₃ (50 ml) was treated with ethyl chloroformate (5 ml) and 15% aq KOH (50 ml). The 2-phase reaction mixt was shaken for 24 hr. Five successive 1-ml portions of ethyl chloroformate were added to the reaction mixt at 1-hr intervals, and KOH soln was added when necessary to maintain a pH > 10. At the end of 24 hr, the CHCl₃ layer was sepd and extd with 1 N HCl. The CHCl₃ was removed in vacuo to afford an oil (3.2 g) whose ir spectrum included characteristic absorptions at 1690 cm⁻¹ (C=O of N-CO₂Et) and 1740 cm⁻¹ (C=O of O⁶-CO₂Et). This was treated with a mixt of MeOH (90 ml) and 10% aq K₂CO₃(10 ml) for 2 hr, concd in vacuo to remove MeOH, extd with Et₂O, and the Et₂O was extd with 1 N HCl. Removal of Et₂O afforded 2.5 g of an oily uncrystallizable material. A soln of 2.0 g of the oil in 95% EtOH (80 ml) was treated with 50% aq KOH (20 ml) and refluxed under N_2 for 24 hr. The soln was dild with H₂O (20 ml) and the EtOH was removed under reduced pressure. The aq acid soln was basified and extd with Et₂O. The Et₂O was removed under reduced pressure to give 0.7 g (yield, 43%) of 7: mp 183-185° (reported mp 185°); 7 · HCl, mp 309-311° dec (reported mp 309° dec). 19

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Synthesis of 4(3H)-Pteridinones

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The sedative-hypnotic activity of some 4(3H)-quinazolones¹ prompted us to synthesize a number of the isosteric 4(3H)-pteridinones and to investigate their hypnotic and

sedative activities. The existing literature gives only a few examples of preparation of 3-alkyl or 3-aryl substituted 4(3H)-pteridinones²⁻⁴ and no data at all on their pharmacology. The synthesis of the title compounds involved the intermediate 3-aminopyrazinecarboxamides, described in Table I, which were obtained, in good yield, from 3-aminopyrazinoic acid (I),⁵ via the mixed anhydride (II)⁶ and reaction of the latter with appropriate amines (R_2NH_2).

$$R_{1} \xrightarrow{N} COOH + CICOOC_{2}H_{5} \xrightarrow{(C_{4}H_{9})_{3}N}$$

$$I$$

$$R_{1} \xrightarrow{N} COOCOOC_{2}H_{5} \xrightarrow{R_{2}NH_{2}} R_{1} \xrightarrow{N} CONHR_{2}$$

$$II \qquad III$$

$$R_{3}C(OC_{2}H_{5})_{3} \xrightarrow{Ac_{2}O} R_{1} \xrightarrow{N} R_{2}$$

$$IV$$

The 3-aminopyrazinecarboxamides (III) could be cyclized to the desired 4(3H)-pteridinones (IV) by condensation with an ortho ester $R_3C(OC_2H_5)_3$ in Ac_2O solution. The amides (III), in contrast to the 4(3H)-pteridinones (IV), reveal a characteristic fluorescence under uv light, which is helpful for their identification by chromatography.

In preliminary CNS screening the majority of the compounds were found to be without hypnotic or sedative activity. Compds 23, 30, and 31 showed a slight sedative activity at 300 mg/kg (mouse) and with 30, 31, and 38 some analgetic activity was observed at 150-250 mg/kg (mouse; phenylbenzoquinone test) and 150-500 mg/kg (mouse; hotplate test), but all the compounds showed too low a therapeutic index, the LD₅₀ (mg/kg; mouse; Litchfield and Wilcoxon) being 1250 (1042-1500), 575 (483-684), 1220 (1070-1391), and 1750 (1400-2187) for 23, 30, 31, and 38, respectively.

Experimental Section

The melting points of all but four compds 21-47 were taken with a Mettler FP-1 apparatus, all the others with a Büchi apparatus, and are uncorrected. Uv and ir spectra were measured for some typical compds and were as expected. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical value.

3-Aminopyrazinecarboxamides (1-20). A mixt of 5.56 g (0.04 mole) of 3-aminopyrazinoic acid, 7.4 g (0.04 mole) of $\mathrm{Bu_3N}$, and 50 ml of dioxane was stirred at room temp until a clear soln resulted. This soln was cooled to 7-8° and 4 ml (0.04 mole) of EtOCOCl was added dropwise, keeping the temp at $11-12^\circ$. After cooling again to 7-8°, 0.04 mole of the appropriate amine hydrochloride was added, and the reaction was allowed to proceed at room temp for 3 hr. The solvent was removed on a rotatory evaporator under reduced pressure and the residue was triturated for 30 min with 50 ml of $\mathrm{H_2O}$, filtered, dried, and recrystd. Recrystn solvents and physical data are given in Table 1

4(3H)-Pteridinones (21-47). A mixt of 0.01 mole of III, 25 ml of ortho ester, and 20-30 ml of Ac₂O was refluxed for 5 hr and then concd on a rotatory evaporator at room temp *in vacuo*. The residue was triturated with 20 ml of EtOH, and, after evapn of the solvent, washed with Et₂O, filtered, dried, and recrystd. Recrystn solvents and physical data are given in Table II. For 23 the reaction was carried out in anhyd HCO₂H, and for 24 in 1:1 anhyd HCO₂H-Ac₂O

Table I. 3-Aminopyrazinecarboxamides (III)

No.	R ₂	R ₁	Formula	Mp, °C	Crystn solvent	Anal.
			$R_1 \searrow N \searrow CONH$	IR ₂		
			101	-		
			N NH ₂			
1	CH ₃	Н	C ₆ H ₈ N ₄ O	134	33% EtOH	C, H, N
2	CH(CH ₃) ₂	H	C ₈ H ₁₂ N ₄ O·HCl·H ₂ O	134	EtOH	C, H, Cl, N, H, C
3	Cyclo-C.H.	H	$C_{10}H_{14}N_{4}O$	73	EtOH-H ₂ O	C, H, N
4	Cyclo-C ₆ H ₁₁	H	$C_{11}H_{16}N_4O$	107-108	33% EtOH	C, H, N
5	C ₆ H ₅	H	$C_{11}H_{10}N_4O$	105	50% EtOH	C, H, N
6	o-ČH₃C₅H₄	H	$C_{12}H_{12}N_4O$	136	50% EtOH	C, H, N
7	m-CHઁ₃Č₄H₄	H	$C_{12}H_{12}N_4O$	112	50% EtOH	C, H, N
8 9	p-CH₃C₅H₄	H	$C_{12}H_{12}N_4O$	158	EtOH	C, H, N
9	o-(i-Pr)C ₆ H ₄	H	$C_{14}H_{16}N_4O$	154	95% EtOH	C, H, N
10	p- $(i$ -Pr $)$ C ₆ H ₄	H	$C_{14}H_{16}N_4O$	125	95% EtOH	C, H, N
11	o-EtOC₅H₄	H	$C_{13}H_{14}N_4O_2$	134-135	65% EtOH	C, H, N
12	m-EtOC ₆ H₄	H	$C_{13}H_{14}N_4O_2$	104-105	65% EtOH	C, H, N
13	p-EtOC ₅ H ₄	H	$C_{13}H_{14}N_4O_2$	139-140	65% EtOH	C, H, N
14	p-BuOC ₆ H ₄	H	$C_{15}H_{18}N_4O_2$	134	EtOH	C, H, N
15	p-C ₆ H ₄ O(CH ₂) ₂ C ₆ H ₅	H H	$C_{19}H_{18}N_4O_2$	180	n-BuOH	C, H, N
16	$CH(C_6H_5)CH_2(C_6H_5)$	H	$C_{19}H_{18}N_{4}O$	136	95% EtOH	C, H, N
17	p-ClC ₆ H ₄	H	C, H, CIN, O	190-191	Dioxane-H ₂ O	C, H, Cl, N
18	3-Pyridyl	H	$C_{10}^{\uparrow}H_{9}N_{5}O$	155-156	EtOH	C, H, N
19	C₅H˙₅	Br	C ₁₁ H ₂ BrN ₄ O	157	95% EtOH	C, H, Br, N
20	o-CH₃C₅H₄	Br	$C_{12}^{11}H_{11}^{1}BrN_{4}^{1}O$	176	95% EtOH	C, H, Br, N

Table II. 4(3H)-Pteridinones (IV)

No.	R ₂	R ₃	R ₁	Formula	Mp, °C	Crystn solvent ^a	Anal.
				O II			
			R_{i}				
			14	Y N N			
				$N \sim N \sim R_3$			
1	CH ₃	H	H	C7H6N4O	294.5-296.5	MeCN	C, H, N
2	CH(CH ₃) ₂	H	H	C ₂ H ₁₀ N ₄ O	204.5-207	EtOH	C, H, N
3	o-CH ₃ C ₆ H ₄	H H	H H	$C_{13}H_{10}N_4O$	166.5-172.5	C ₆ H ₆ -P	C, H, N
4	m-CH ₃ C ₆ H ₄			$C_{13}H_{10}N_4O$	235-236.5 239-241.5	CHCl₃-Et₂O	C, H, N
5	p-C ₂ H ₅ OC ₆ H ₄	.H H	H Br	$C_{14}H_{12}N_4O_2$	$^{239-241.5}_{\sim 147^d}$	MeCN EtOH	H, N, C ^b
6	<i>о</i> -СН₃С ₆ Н₄ Н	CH,	H	C ₁₃ H ₉ BrN ₄ O	>299	MeCN	C, H, Br, N C, H, N
7 8	CH(CH ₃) ₂	CH ₃	H	C ₇ H ₆ N ₄ O C ₁₀ H ₁₂ N ₄ O	202-205.5	THF	C, H, N C, H, N
)	Cyclo-C ₅ H ₆	CH ₃	H	$C_{10}H_{12}N_4O$ $C_{12}H_{14}N_4O$	164.5-167	n-Bu₂O	C, H, N
Ó	Cyclo-C ₆ H ₁₁	CH ₃	H	$C_{13}H_{16}N_4O$	203-204.5	C ₆ H ₆ -P	C, H, N
1	o-CH ₃ C ₆ H ₄	CH ₃	H	$C_{14}H_{12}N_4O$	188.5-190.5	C_6H_6-P	C, H, N
2	m-CH ₃ C ₆ H ₄	CH ₃	H	$C_{14}H_{12}N_4O$	201-204.5	EtOH	C, H, N
3	p-CH ₃ C ₆ H ₄	CH ₃	H	$C_{14}H_{12}N_4O$	226-227.5	EtOH	C, H, N
4	o-(i-Pr)C ₆ H ₄	CH ₃	H	$C_{16}H_{16}N_4O$	179.5-182.5	EtOH	C, H, N
5	p- $(i$ -Pr $)$ C ₆ H ₄	CH ₃	H	$C_{16}H_{16}N_{4}O$	177.5-180.5	EtOH-i-Pr ₂ O	C, H, N
6	o-EtOC ₆ H ₄	CH ₃	H	$C_{15}H_{14}N_{4}O_{2}$	240.5-243.5	MeOEtOH	C, H, N
7	m -EtOC $_6$ H $_4$	CH ₃	H	$C_{15}H_{14}N_4O_2$	148^d	EtOH	C, H, N
8	p-EtOC ₆ H ₄	CH ₃	H	$C_{15}H_{14}N_4O_2$	196-198.5	EtOH	H, N, C ^c
9	p-C ₄ H ₉ OC ₆ H ₄	CH ₃	H	$C_{17}H_{19}N_4O_2$	$\sim 144^{d}$	EtOH	C, H, N
0	C ₆ H ₅ (CH ₂) ₂ OC ₆ H ₄	CH ₃	H	$C_{21}H_{18}N_4O_2$	155-159.5	EtOH	C, H, N
	C_6H_5 C_6H_5						
1	-CHCH ₂	CH ₃	Н	$C_{21}H_{18}N_4O$	198.5-204	C ₆ H ₆ -P	C, H, N
2	3-Py	CH ₃	H	C ₁₂ H ₉ N ₅ O	200.5 dec	EtOH	C, H, N
13	C ₆ H́ ₅	CH,	Br	C ₁₃ H ₀ BrN ₄ O	~195 ^d	C ₆ H ₆ -P	C, H, Br, N
4	Cyclo-C ₆ H ₁₁	C ₂ H ₅	H	$C_{14}H_{18}N_4O$	147.5-150.5	EtOH	C, H, N
5	p- $(i$ -Pr $)$ C ₆ H ₄	C_2H_5	H	$C_{17}H_{18}N_4O$	160-163	EtOH	C, H, N
6	p-EtOC ₄ H ₄	C_2H_5	H	$C_{16}H_{16}N_4O_2$	165.5-167.5	EtOH	C, H, N
17	p-EtOC ₆ H ₄	CH₂C₅H₅	H	$C_{21}H_{18}N_4O_2$	117-121.5	EtOH-Et ₂ O	C, H, N

^aP = petr ether. ^bCalcd, 62.67; found, 62.15. ^cCalcd, 63.82; found, 64.35. ^dBüchi apparatus.

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